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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/063,236	04/02/2002	Richard M. Webber	H-317 5209	
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DAVID J C	OLE	MARTINEZ, JOSEPH P		
E INK CORPORATION 733 CONCORD AVE			ART UNIT	PAPER NUMBER
CAMBRIDGE, MA 02138-1002			2873	
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Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)			
Office Action Commons	10/063,236	WEBBER, RICHARD M.			
Office Action Summary	Examiner	Art Unit			
	Joseph P. Martinez	2873			
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address			
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a reply If NO period for reply is specified above, the maximum statutory period we Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	within the statutory minimum of thirty (30) days ill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONEI	nely filed s will be considered timely. the mailing date of this communication. D (35 U.S.C. § 133).			
Status					
1) Responsive to communication(s) filed on 19 Ma	ay 2005.				
☐ This action is FINAL. 2b) ☐ This action is non-final.					
3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is					
closed in accordance with the practice under E	x parte Quayle, 1935 C.D. 11, 45	53 O.G. 213.			
Disposition of Claims					
4) ⊠ Claim(s) 1-28 is/are pending in the application. 4a) Of the above claim(s) is/are withdraw 5) □ Claim(s) is/are allowed. 6) ⊠ Claim(s) 1-28 is/are rejected. 7) □ Claim(s) is/are objected to. 8) □ Claim(s) are subject to restriction and/or	vn from consideration.				
Application Papers					
9) ☐ The specification is objected to by the Examiner 10) ☑ The drawing(s) filed on 02 April 2002 is/are: a) Applicant may not request that any objection to the ore Replacement drawing sheet(s) including the correction 11) ☐ The oath or declaration is objected to by the Examiner	☑ accepted or b) ☐ objected to be drawing(s) be held in abeyance. See ton is required if the drawing(s) is obj	e 37 CFR 1.85(a). jected to. See 37 CFR 1.121(d).			
Priority under 35 U.S.C. § 119	•				
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 					
Attachment(s)					
Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:				
S. Patent and Trademark Office					

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DETAILED ACTION

Response to Arguments

Applicant's arguments, see p. 1-2, filed 5-19-05, with respect to the use of prior art Herb et al. (6693620) have been fully considered and are persuasive. Therefore, the rejection has been withdrawn. However, upon further consideration, a new ground(s) of rejection is made in view of Albert et al. (6017584).

Applicant's arguments with respect to prior art Albert et al. (6017584) have been considered but are most in view of the new ground(s) of rejection.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-28 are rejected under 35 U.S.C. 103(a) as being unpatentable over Albert et al. (6017584).

Re claims 1-10, Albert et al. teaches for example in fig. 1, an electrophoretic medium comprising a plurality of particles (15) suspended in a suspending fluid (17), the particles being capable of moving through the fluid upon application of an electric field to the medium (col. 2, In. 48-50), the fluid having dissolved or dispersed therein a

polymer (polyisobutylene succinimide as a charge control agent or particle stabilizer, col. 19, ln. 17) and the polymer is polyisobutylene (col. 19, ln. 17).

But, Albert et al. fails to explicitly teach the polymer being essentially non-absorbing on the particles; a number average molecular weight in excess of about 20,000; a number average molecular weight in excess of about 100,000; a number average molecular weight in the range of about 1 50,000 to about 3,000,000; a weight average molecular weight in excess of about 1 00,000; a weight average molecular weight in the range of about 300,000 to about 3,000,000; the polymer has a polydispersity index not greater than about 2; the polymer is a hydrocarbon polymer essentially free from aromatic groups; the polymer is a polyolefin; and the polyisobutylene has a viscosity average molecular weight in the range of about 200,000 to 1,200,000 g/mole.

However, it has been held that products of identical chemical composition cannot have mutually exclusive properties (see MPEP § 2112.02). Therefore, if the prior art teaches the identical chemical structure, the properties applicant discloses and/or claims are necessarily present. *In re Spada*, 911 F.2d 705, 709, 15 USPQ2d 1655, 1658.

Furthermore, the office interprets the teachings of the use of polyisobutylene to disclose the polymer is a hydrocarbon polymer essentially free from aromatic groups and the polymer being essentially non-absorbing on the particles since it is property of the claimed composition.

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Still furthermore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to vary the molecular weight of the polymer, since it has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 105 USPQ 233.

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the teachings of Albert et al. to vary the molecular weight of polyisobutylene in order to control particle-particle and particle-wall attraction.

Re claim 24, Albert et al. teaches for example in fig. 1, an electrophoretic medium comprising a plurality of particles (15) suspended in a suspending fluid (17), the particles being capable of moving through the fluid upon application of an electric field to the medium (col. 2, ln. 48-50), the fluid having dissolved or dispersed therein a polymer (polyisobutylene succinimide as a charge control agent or particle stabilizer, col. 19, ln. 17) and the polymer is polyisobutylene (col. 19, ln. 17).

But, Albert et al. fails to explicitly teach the polyisobutylene has a viscosity average molecular weight in the range of about 400,000 to 1,200,000 g/mole and the polymer is present in an amount of from about 0.5 to about 2.5 percent by weight of the suspending fluid.

However, it has been held that products of identical chemical composition cannot have mutually exclusive properties (see MPEP § 2112.02). Therefore, if the prior art

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teaches the identical chemical structure, the properties applicant discloses and/or claims are necessarily present. *In re Spada*, 911 F.2d 705, 709, 15 USPQ2d 1655, 1658.

Furthermore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to vary the molecular weight and concentration of the polymer, since it has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 105 USPQ 233.

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the teachings of Albert et al. to vary the molecular weight and concentration of polyisobutylene in order to control particle-particle and particle-wall attraction.

Re claim 28, Albert et al. teaches for example in fig. 1, an electrophoretic medium comprising a plurality of particles (15) suspended in a suspending fluid (17), the particles being capable of moving through the fluid upon application of an electric field to the medium (col. 2, ln. 48-50), the fluid having dissolved or dispersed therein a polymer (polyisobutylene succinimide as a charge control agent or particle stabilizer, col. 19, ln. 17) and the polymer is polyisobutylene (col. 19, ln. 17).

But, Albert et al. fails to explicitly teach a polymer having an intrinsic viscosity of n in the suspending fluid and being substantially free from ionic or ionizable groups in

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the suspending fluid, the polymer being present in the suspending fluid in a concentration of from about $0.5 [\eta]^{-1}$ to about $2.0 [\eta]^{-1}$.

However, it has been held that products of identical chemical composition cannot have mutually exclusive properties (see MPEP § 2112.02). Therefore, if the prior art teaches the identical chemical structure, the properties applicant discloses and/or claims are necessarily present. *In re Spada*, 911 F.2d 705, 709, 15 USPQ2d 1655, 1658.

Furthermore, the office interprets the teachings of the use of polyisobutylene to disclose the polymer being substantially free from ionic or ionizable groups and having an intrinsic viscosity since these are properties of the claimed composition.

Furthermore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to vary the concentration of the polymer, since it has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*. 105 USPQ 233.

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the teachings of Albert et al. to vary the concentration of the polymer in order to control particle-particle and particle-wall attraction.

Re claim 11, Albert et al. further teaches for example, the polymer is a polysiloxane (col. 16, In. 50-52).

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Re claims 12 and 13, Albert et al. further teaches for example, the suspending fluid is an aliphatic hydrocarbon (col. 16, ln. 45) or a mixture (col. 16, ln. 24-26) of an aliphatic hydrocarbon and a halogenated hydrocarbon (col. 16, ln. 55).

Re claim 14, supra claim 1.

But, Albert et al. fails to explicitly teach the suspending fluid and polymer are such that the scaling exponent for variation of intrinsic velocity with molecular weight falls in the range of about 0.55 to about 0.8.

However, it has been held that products of identical chemical composition cannot have mutually exclusive properties (see MPEP § 2112.02). Therefore, if the prior art teaches the identical chemical structure, the properties applicant discloses and/or claims are necessarily present. *In re Spada*, 911 F.2d 705, 709, 15 USPQ2d 1655, 1658.

Furthermore, the office interprets the intrinsic viscosity is intrinsic to the composition and that the prior art teaches the claimed composition. Furthermore, the prior art teaches varying the molecular weight. The office interprets the teachings of Albert et al. to disclose varying the molecular weight of the polymer in regards to the suspending fluid, which has an intrinsic viscosity, to therefore teach a scaling exponent.

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to have the scaling exponent for variation of intrinsic velocity with molecular weight fall within the range of about 0.55 to about 0.8, since it

has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re*Aller, 105 USPQ 233.

Re claims 15 and 16, supra claim 1.

But, Albert et al. fails to explicitly teach the polymer is present in an amount of from about 0.25 to about 2.5 percent by weight or from about 1 to about 2 percent by weight of the suspending fluid.

However, it would have been obvious to one of ordinary skill in the art at the time the invention was made to vary the concentration of the polymer, since it has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 105 USPQ 233.

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the teachings of Albert et al. to vary the concentration of the polymer in order to control particle-particle and particle-wall attraction.

Re claim 17, Albert et al. further teaches for example in fig. 1, an electrophoretic medium is encapsulated (col. 7, ln. 15), with the suspending fluid (17) and particles (15) being retained within a plurality of capsules (13).

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Re claim 18, Albert et al. further teaches for example, an electrophoretic medium is of a two-phase type, the suspending fluid and particles forming a discontinuous phase comprising a plurality of droplets, the droplets being surrounded by a continuous phase (col. 2, In. 24-31).

Re claim 19, supra claim 18.

But, Albert et al. fails to explicitly teach said droplets comprise at least about 40 percent by volume of the electrophoretic medium.

However, it would have been obvious to one of ordinary skill in the art at the time the invention was made to vary the concentration of the electrophoretic medium, since it has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 105 USPQ 233.

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the teachings of Albert et al. to vary the concentration of the polymer in order to provide a coherent display.

Re claims 20 and 21, Albert et al. further teaches for example in fig. 1 and 2, an electrophoretic medium is of the dual particle type having two different types of particles (10 and 12) having different electrophoretic mobilities (col. 3, In. 8-15) or the suspending fluid (17) and the two types of particles (10 and 12) being retained within a plurality of capsules (14).

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Re claims 22 and 23, supra claim 1.

But, Albert et al. fail to explicitly teach an image stability of at least about 1,000 seconds or about 10,000 seconds.

However, Albert et al. teaches for example, "whether or not an encapsulated electrophoretic display is bistable, and its degree of bistability, can be controlled through appropriate chemical modification of the electrophoretic particles, the suspending fluid, the capsule, and binder materials" (col. 6, In. 43-47) and further teaches controlling the appropriate chemical modification of the electrophoretic particles (col. 12-15), the suspending fluid (col. 15-17), the capsule (col. 19 and 20), and binder materials (col. 22 and 23) and still further suggests stability times of hours or days (col. 6, In. 32-35).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to teach an image stability of at least about 1,000 seconds or about 10,000 seconds, since it has been held that where the general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller*, 105 USPQ 233.

Re claim 25, Albert et al. further teaches for example in fig. 1, at least one electrode (16 and 16') arranged adjacent the medium and capable of applying an electric field to the medium (col. 7, ln.26-30).

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Re claim 26, Albert et al. further teaches for example in fig. 1, two electrodes (16 and 16') disposed on opposed sides of the electrophoretic medium, at least one of the electrodes being substantially transparent (col. 12, ln. 31-34) such that the electrophoretic medium can be viewed through the substantially transparent electrode (col. 12, ln. 39-40).

Re claim 27, Albert et al. further teaches for example in fig. 1, the suspending fluid (17) and particles (15) being retained within a plurality of capsules (13), the capsules being retained within a solid binder (11), and the electrode (16 and 16') being secured to the binder (col. 34-38).

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Joseph P. Martinez whose telephone number is 571-272-2335. The examiner can normally be reached on M-F 7:00 AM to 3:30 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Georgia Y. Epps can be reached on 571-272-2328. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

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JPM 7-28-05

Georgia Epps
Georgia Epps
Supervisory Patent Examiner
Technology Center 2800